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(54) PHOTOVOLTAIC CELL WITH THIAZOLE-CONTAINING POLYMER

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CPC .. C08G 61/123; C08G 75/06; H01L 51/0036; H01L 51/4226; Y02E 10/549 USPC 528/289, 27, 226, 377, 380; 136/263 See application file for complete search history.

(56) References Cited

U.S. PATENT DOCUMENTS

4,292,092 A	9/1981	Hanak
4,416,959 A	11/1983	Skotheim
4,574,160 A	3/1986	Cull et al.
4,639,328 A	1/1987	Krause et al.
4,686,323 A	8/1987	Biter et al.
4,746,618 A	5/1988	Nath et al.
4,795,687 A	1/1989	Suzuki et al.
4,913,744 A	4/1990	Hoegl et al.
4,948,436 A	8/1990	Juergens
5,071,490 A	12/1991	Yokota et al.
5,221,363 A	6/1993	Gillard
5,274,058 A	12/1993	Ferraris et al.
5,510,438 A	4/1996	Ferraris et al.

5,536,808	A	7/1996	Curtis et al.
5,708,130	A	1/1998	Woo et al.
6,109,330	Α	8/2000	Crisp
6,132,585	A	10/2000	Midorikawa et al
6,188,175	B1	2/2001	May et al.
6,198,091	B1	3/2001	Forrest et al.
6,198,092	В1	3/2001	Bulovic et al.
6,239,355	В1	5/2001	Salafsky
6,278,055	B1	8/2001	Forrest et al.
6,297,495	В1	10/2001	Bulovic et al.
6,333,458	В1	12/2001	Forrest et al.
6,352,777	В1	3/2002	Bulovic et al.
6,353,083	В1	3/2002	Inbasekaran et al.
6,399,224	В1	6/2002	Li
6,440,769	B2	8/2002	Peumans et al.
6,451,415	В1	9/2002	Forrest et al.
6,464,762	B1	10/2002	Arao
6,580,027	B2	6/2003	Forrest et al.
6,657,378	B2	12/2003	Forrest et al.
6,772,817	B2	8/2004	Lai et al.
6,818,260	B2	11/2004	Farrand et al.
6,864,333	B2	3/2005	Dall'Occo et al.
7,095,044	B2	8/2006	Brown et al.
7,105,237	B2	9/2006	Sotzing
7,196,366	B2	3/2007	Forrest et al.
7,309,833	B2	12/2007	Robeson et al.
7,368,510	B2	5/2008	Lee et al.
7,405,775	B2	7/2008	Nilsson
2002/0040728	A1	4/2002	Yoshikawa
2002/0050289	Al	5/2002	Wada et al.
2002/0105005	A1	8/2002	Seo et al.
2003/0008172	A1	1/2003	Leclerc et al.
		(Con:	tinued)
		n .on	concect t

(Continued)

FOREIGN PATENT DOCUMENTS

DE 31 21 350 7/1982 DE 103 26 547 1/2005 (Continued)

OTHER PUBLICATIONS

Kong et al. "Single-Walled Carbon Nanotube Gold Nanohybrids: Application in Highly Effective Transparent and Conductive Films", J. Phys. Chem. C 2007, 111, p. 8377-8382.

Peng et al. "Synthesis and electroluminescent properties of copolymers based on fluorene and 2,5-di(2-hexyloxyphenyl)thiazolothiazole" Macromolecules 38, 7292-7298 (2005).

Ando et al. "synthesis, physical properties, and field-effect transistors of novel thiophene/thiazolothiazole co-loligomers" J. Mater. Chem. 14, 1787-1790(2004).

Gao et al. "Efficient photodetectors and photovoltaic cells from composites of fullerenes and conjugated polymers: photimduced electron transfer" Synthetic Metal 84, 979-980(1997).

(Continued)

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(57) ABSTRACT

Photovoltaic cells with thiazole-containing polymers, as well as related components, systems, and methods, are disclosed.

18 Claims, 1 Drawing Sheet

WO WO 2007/145482 12/2007 (56)References Cited WO WO 2008/000664 1/2008 WO WO 2010/008672 1/2010 U.S. PATENT DOCUMENTS OTHER PUBLICATIONS 2003/0023029 A1 1/2003 Wang et al. Barbarella et al., "Oligothiophene-S,S dioxides: a New Class of 2003/0036612 A1 2/2003 Nifant'ev et al. 2003/0042471 A1 Thiophene based Materials", Journal of Organic Chemistry, Optical 3/2003 Giles et al. 2003/0102024 A1 6/2003 Zeira Materials, 9 (1998) 43-45. 2003/0127967 A1 7/2003 Tsutsui et al. European Search Report dated Feb. 24, 2011, Application No. 2003/0159729 A1 8/2003 Shaheen et al. 10177632.6. 2003/0175411 A1 9/2003 Kodas et al International Preliminary Report on Patentability dated Feb. 8, 2011 2003/0188777 A1 10/2003 Gaudiana et al. issued in PCT/US2009/049701. 2003/0189402 A1 10/2003 Gaudiana et al. Jayakanna, et al., "Synthesis and Structure-Property Relationship of 2003/0230335 A1 12/2003 Tsukahara New Donor-Acceptor-Type Conjugated Monomers on the Basis of 2004/0118448 A1 6/2004 Scher et al. Thiophene and Benzothiadiazole", Journal of Polymer Science Part 2004/0187911 A1 9/2004 Gaudiana et al. A: Polymer Chemistry, vol. 40, 251-261 (2002). 2004/0192871 A1 9/2004 Wang et al. Kunai et al., "Synthesis and properties of silicon-bridged 2004/0201018 A1 10/2004 Yamahara et al. bithiophenes and application to EL devices," Synthetic Metals, 2004/0256615 A1 12/2004 Sirringhaus et al. Elsevier Sequoia Lausanne, CH, vol. 137, No. 1-3, Jan. 1, 2003, pp. 2/2005 2005/0022865 A1 Robeson et al. 2005/0124784 A1 6/2005 Sotzing 1007-1008 2005/0145972 A1 7/2005 Fukuda et al. Supplemental European Search Report dated Mar. 4, 2011, Applica-2005/0224905 A1 10/2005 Forrest et al. tion No. 07872313.7. 2006/0022192 A1 2/2006 Brabec et al. Supplemental European Search Report dated Mar. 4, 2011, Applica-2006/0027834 A1 2/2006 Forrest et al. tion No. 06787321.6. 2006/0076050 A1 4/2006 Williams et al. Ahmed, et al., "Benzobisthiazole-Thiophene Copolymer Semicon-2006/0141662 A1 6/2006 Brabec et al. ductors: Synthesis, Enhanced Stability, Field-Effect Transistors, and 2006/0155106 A1 7/2006 Roberts et al. Efficient Solar Cells", Macromolecules 2009, 42, 8615-8618. 2007/0014939 A1 1/2007 Gaudiana et al. Ando, et al., "Characterization and Field-Effect Transistor Perfor-2007/0017571 A1 1/2007 Gaudiana et al. mance of Heterocyclic Oligomers Containing a Thiazolothiazole 2007/0020526 A1 1/2007 Gaudiana et al. Unit", Chemistry Letter vol. 33, No. 9, 1170-1171 (2004). 2007/0120045 A1 5/2007 Yokoyama 2007/0158620 A1 7/2007 Gaudiana et al. Chen, et al., "Polymer solar cells with enhanced open-circuit voltage 2007/0181179 A1* 8/2007 Brabec et al. 136/263 and efficiency", Nature Photonics, vol. 3, 649-653, Nov. 2009. 2007/0193621 A1 8/2007 Brabec et al. European Search Report dated Nov. 10, 2010, Application No. 2007/0289626 A1* 12/2007 Brabec et al. 136/263 10177632.6. 2008/0053518 A1 3/2008 Change et al. Hou, et al., "Synthesis, Characterization, and Photovoltaic Properties 2008/0264488 A1 10/2008 Balasubramanian of a Low Band Gap Polymer Based on Silole-Containing Polythiophenes and 2, 1, 3-Benzothiadiazole", J. Am. Chem. Soc. FOREIGN PATENT DOCUMENTS 2008, 130, 16144-16145. Hou, et al., "Bandgap and Molecular Energy Level Control of Con-DE 69330835 2/2007 jugated Polymer Photovoltaic Materials Based on Benzo[1,2-b:4,5-0 251 598 1/1988EΡ b']dithiophene", Macromolecules 2008, 41, 6012-6018. EΡ 1 065 725 1/2001 ΕP 1318553 6/2003 International Search Report dated Jan. 27, 2011, PCT/US2010/ 1507298 2/2005 EP 026222. ΕP 1562154 8/2005 Li, et al "High-efficiency solution processable polymer photovoltaic JP JP 04-192376 7/1992 cells by self-organization of polymer blends", Nature Materials vol. 2001 060707 3/2001 4, 864-868, Nov. 2005. JP JP 2002-141524 5/2002 Liu et al., "Efficient green-light-emitting diodes from silole-contain-03-073382 3/2003 ing copolymers", 15, 3496-3500 (2003). Maksudul, et al., "Polybenzobisazoles Are Efficient Electron Trans-JР 2003-515933 5/2003 JР 2003 264085 9/2003 JР 04-534863 6/2004 port Materials for Improving the Performance and Stability of Poly-JΡ 05-011841 1/2005 mer Light-Emitting Diodes", Chem. Mater. 2002, 14, 4775-4780. JΡ 06-166746 6/2006 Mike, et al., "Facile Synthesis of 2, 6-Disubstituted JР 06-278682 10/2006 Benzobisthiazoles: Functional Monomers for the Design of Organic KR 100685796 2/2007 Semiconductors", J. Org. Chem. 2010, 75, 495-497. WO 93/05077 3/1993 WO Mike, et al., An Efficient Synthesis of 2, 6-Disubstituted WO WO 94/14199 6/1994 Benzobisoxazoles: New Building Blocks for Organic Semiconduc-WO WO 97/05184 2/1997 tors, Org. Lett., vol. 10, No. 21, 4915-4918, 2008. WO WO 00/11725 3/2000 Hiramoto et al., "Design of nanostructures for photoelectric conver-WO WO 00/22682 4/2000 sion using an organic vertical superlattice," Appl. Phys. Letters, WO WO 00/46321 8/2000 WO WO 01/39276 5/2001 88:213105, 2006. WO WO 02/059121 A1 8/2002 Osahen, et al., "Electroactive and Photoactive Rod-Coil Copolymers: WO WO 02/101838 12/2002 Design, Synthesis, and Supramolecular Regulation of Photophysical WO WO 03/065462 8/2003 Properties", J. Am. Chem. Soc. 1995, 117, 7389-7398. WO WO 2004/042824 5/2004 Osaka, et al., "Novel Thiophene-Thiazolothiazole Copolymers for WO WO 2004/051756 6/2004 Organic Field-Effect Transistors", Adv. Mater. 2007, 19, 4160-4165. WO WO 2005/092947 10/2005 Osaka, et al., "High-Lamellar Ordering and Amorphous-Like it-Net-WO WO 2005/111045 11/2005 work in Short-Chain Thiazolothiazole-Thiophene Copolymers Lead WO WO 2007/011739 1/2007 to High Mobilitiesz", J. Am. Chem. Soc. 2009, 131, 2521-2529. WO WO 2007022226 2/2007 Takamoto et al., "Two-Terminal Monolithic Ino.5Gao.5P/GaAs Tan-WO 2007/076427 7/2007 WO dem Solar Cells with a High Conversion Efficiency of Over 30%," WO WO 2007/121252 10/2007

Jap. J. Appl. Physics, 36(10):6215-6220.

WO

WO 2007/133705

11/2007

(56) References Cited

OTHER PUBLICATIONS

Umeda et al., "Fabrication of Interpenetrating Semilayered Structure of Conducting Polymer and Fullerene by Solvent Corrosion Method and Its Photovoltaic Properties," Jap. J. Appl. Physics, 44(6A):4155-4160, 2005.

Xue, et al., "Asymmetric tandem organic photovoltaic cells with hybrid planar-mixed molecular heterojunctions," Appl. Phys. Letters, 85(23):5757-5759, 2004.

Yakimov et al., "High photovoltage multiple-heterojunction organic solar cells incorporating interfacial metallic nanoclusters," Applied Physics Letters, vol. 80, No. 9, Mar. 2002.

Yamaguchi et al., "Multi-junction III-V solar cells: current status and future potential," Solar Energy, 79(1):78-85, 2005.

Peet et al., "Efficiency enhancement in low-bandgap polymer solar cells by processing with alkane dithiols," Nat. Mat., 6:497-500, 2007. Shrotriya, V., "Plastic Solar Cells", Organic Semiconductor Conference, 1-33 (2009).

So, et al "Synthesis and Photophysical Properties of Some Benzoxazole and Benzothiazole Compounds", Macromolecules 1996, 29, 2784-2795.

Zhang, et al., "Polymer Solar Cells Based on a Low-Bandgap Fluorene Copolymer and a Fullerene Derivative with Photocurrent Extended to 850 nm", Adv. Funct. Mater. 2005, No. 15, 745-750.

Swager et al., "The molecular wire approach to sensory signal amplifiers," Acc. Chem. Res., 1998, 31, 201-207.

Van Duren, "R.A.J. Low-bandgap Polymer Photovoltaic Cells," Synthetic Metals, 121:1587-1588, 2001.

Zhou et al., "Fluorene based low band gap copolymers for photovoltaic devices," Applied Physics Letters, vol. 84, No. 10, 1653-1655, Mar. 8, 2004.

A. C. Arias, et al., Vertically segregated polymer-blend photovoltaic thin-film structures through surface-mediated solution processing, applied physics letters, vol. 80, No. 10 Mar. 11, 2002.

Anthony M Higgins, et al., "Surface segregation and self-stratification in blends of spin-cast polyfluorene derivatives", J. Phys.: Condens. Matter 17 (2005) 1319-1328.

Hwajeong Kim, et al., "Distinct Annealing Temperature in Polymer:Fullerene:Polymer Ternary Blend Solar Cells", J. Phys. Chem. C, 2009, 113 (4), 1620-1623 DOI: 10.1021/jp809589n, Publication Date (Web): Jan. 5, 2009.

Gadisa et al., "Improvements of fill factor in solar cells based on blends of polyfluorene copolymers as electron donors," Thin Solid Films, 515:3126-3131, (2007).

Kim et al., "Influence of electron-donating polymer addition on the performance of polymer solar cells," J. Phys. D: Appl. Phys., 41:1-5, (2008).

Winder et al., "Sensitization of low bandgap polymer bulk heterojunction solar cells," Thin Solid Films, 403-404:373-379, (2002).

Wong et al., "Efficient photovoltaic cells with wide photosensitization range fabricated from rhenium benzathiazole complexes," App. Phys. Letters, 90:1-3, (2007).

Hu and Matsumura, "Structure and thickness dependence of p-n heterojunction solar cells based on copper phthalocyanine and perylene pigments", Institute of Physics Publishing Journal of Physics D: Applied Physics, J. Phys. D: Appl. Phys., 37:1434-1438 (2004).

Peumans et al., "Efficient photon harvesting at high optical intensities in ultrathin organic double-heterostructure photovoltaic diodes", American Institute of Physics, Applied Physics Letter, 76(19):2650-2652 (2000).

Roman et al., "High Quantum Efficiency Polythiophene/C60 Photodiodes," Adv. Mat., 10:774-777, 1998.

Jespersen et al., "The electronic states of polyfluorene copolymers with alternating donor-acceptor units," J. Chem. Phy., 121(24):12613-12617, 2004.

Sensfuss et al., Characterisation of potential donor acceptor pairs for polymer solar cells by ESR, optical and electrochemical investigations, vol. 5215, Proc. of SPIE129-140, 2004.

Baumgartner, "π-Conjugated Heterocyclic fused Bithiophene Materials," J. Inorganic and Organometallic Polymers and Materials, 15(4):389-409, 2005.

Coppo et al., "New routes to poly(4,4-dialkylcyclopentadithiophene-2,6-diyls)," J. Mater. Chem., 12:2597-2599, 2002.

Muhlbacher et al., "High Photovoltaic Performance of a Low-Bandgap Polymer," Adv. Mater., 18:2884-2889, 2006.

Usta et al., "Dithienosilole- and Dibenzosilole- Thiophene Copolymers as Semiconductors for Organic Thin-Film Transistors," J. Am. Chem., 128:9034-9035, 2006.

Zhu et al., "Panchromatic Conjugated Polymers Containing Alternation Donor/Acceptor Units for Photovoltaic Applications," Macromolecules, 40:1981-1986, 2007.

Hiramoto et al., "Effect of Thin Gold Interstitial-layer on the Photovoltaic Properties of Tandem Organic Solar Cell," Chemistry Letters, The Chemistry Society of Japan, pp. 327-330, 1990.

Liao et al., "Photovoltaic Dithienosilole-Containing Polymers," Macromoleculs, vol. 40, pp. 9406-9412 (2007).

Niu, Applied Physics Letters, 2003, vol. 82, No. 13, 2163-2165.

Ohshita et al., "Synthesis of Polymers Composed of Alternating Diphenylenedithienosilole and Diethynylenesilyene Units and Their Applications to Hole Transport in Double-Layer EL Devices" Macromolecules, vol. 33, pp. 8890-8893, 2000. Entire document, especially Scheme 1, p. 8890; Table 2.

Plater et al., "Cyclopenta-1,2,3-dithiazoles and related compounds," J. Chem.. Soc., Perkin Trans. 1, Issue 7, 769-774, 1993. Abstract. Asawapirom et al., "Dialkylcyclopentadithiophene Polymers and

Asawapirom et al., "Dialkylcyclopentadithiophene Polymers and Copolymers," Macromolecular Rapid Communications, 22:746-749, 2001.

Bakhshi et al., "Ab initio study of the electronic structures and conduction properties of some donor-acceptor polymers and their copolymers," Journal of Chemical Physics, 119:13159-13169, 2003. Bakhshi et al., "Electronic Structures of Donor-Acceptor Polymers Based on Polythiophene, Polyfuran and Polypyrrole," Molecular Engineering, 6:239-248, 1996.

Bakhshi et al., "Strategies for Molecular Designing of Novel Low-Band-Gap Electrically Conducting Polymers," Applied Biochemistry and Biotechnology, 96:125-133, 2001.

Berlin et al., "Self-assembly of mono- and multilayers of polyconjugated conducting polymers," Macromolecular Rapid Communications, 21:301-318, 2000.

Berlin et al., "Thiophene/Cyclopentadiene Regular Copolymers from Electrochemical Oxidation of Dithienylcyclopentadienes," Macromolecular Chemistry and Physics, 203:1228-1237, 2002.

Coppo et al., "Cyclopentadithiophene based electroactive materials," J Mater. Chem., 15:1123-1133, 2005.

Coppo et al., "Investigation of the electronic properties of cyclopentadithiophene polymers and copolymers," Mat. Res. Soc. Symp. Proc., 771:L4.9.1-L4.9.4, 2003.

Coppo et al., "Synthesis, solid state structure and polymerisation of a fully planar cyclopentadithiophene," Chem. Commun., 2548-2549, 2003

Coppo et al., "Synthetic Routes to Solution-Processable Polycyclopentadithiophenes," Macromolecules, 35:2705-2711, 2003.

Cravino et al., "Double-cable polymers for fullerene based organic optoelectronic applications," J. Mater. Chem., 12:1931-1943, 2002. Cravino et al., "Nanostructured Bipolar Organic Polymers," Encyclopedia of Nanoscience and Nanotechnology, 7:615-628.

Dai, "Conjugated and Fullerene-Containing Polymers for Electronic and Photonic Applications: Advanced Syntheses and Microlithographic Fabrications," Journal of Macromolecular Science, Part C—Reviews in Macromolecular Chemistry and Physics, 39:273-387, 1999.

Huang et al., "A Donor-Acceptor Conducting Copolymer with a Very Low Band Gap and High Intrinsic Conductivity," Chem. Mater., vol. 10, No. 8, pp. 2212-2216, (1998).

Huang et al., "In-situ conductivity of a low band-gap conducting polymer. Measurement of intrinsic Conductivity," Acta Polymerica, vol. 48(10), (1997).

Huang et al., "Oxygen-Modified Poly(4-dicyanomethylene-4H-cyclopenta[2,1-b;3,4-b]dithiophene: A Tunable Low band Gap Polymer," Chem. Mater., vol. 11, No. 6, pp. 1541-1545, (1999).

(56) References Cited

OTHER PUBLICATIONS

Ko et al., "Electrochemistry and electrochromism of a poly(cyclopentadithiophene) derivative with a viologa pendant," Electrochimica Acta. vol. 48, pp. 4127-4135, (2003).

Kunugi et al., "Charge transport in a regioregular poly(3-octylthiophene) film," J Mater. Chem., 10:2673-2677, 2000.

Larmat et al., "Comparative Reactivity of Thiophene and 3,4-(Ethylenedioxy) thiophene as Terminal Electropolymerizable Units in Bis-Heterocycle Arylenes," Journal of Polymer Science: Part A: Polymer Chemistry, 35:3627-3636, 1997.

Prato, "Fullerene Materials," Topics in Current Chemistry, 199:173-187, 1999.

Roncali, "Linear π -conjugated systems derivatized with C60-fullerene as molecular heterojunctions for organic photovoltaics," Chem. Soc. Rev., 34:483-495, 2005.

Roncali, "Linear π-conjugated systems with tailored electronic properties," Annu. Rep. Prog. Chem., Sect. C, 95:47-88, 1999.

Scherf et al., "Semiconducting Polyfluorenes—Towards Reliable Structure—Property Relationships," Advanced Materials, 14:477-487, 2002.

Subramanian et al., "Trends in Geometric and Electronic Properties of Thiophene- and CyclopentadieneBased Polymers," International Journal of Quantum Chemistry, 66:229-240, 1998.

Wang et al., "Conjugated Fluorene and Silole Copolymers: Synthesis, Characterization, Electronic Transition, Light Emission, Photovoltaic Cell, and Field Effect Hole Mobility," Macromolecules, 38:2253-2260, 2005.

Wu et al., "Low band gap-conjugated polymer derivatives," Synthetic Metals. vol. 155, pp. 618-622 (2005).

Yonezawa et al., "Preparation and properties of oligomers containing cyclopentadithiophene," Synthetic Metals, 119:99-100, 2001.

Yoon et al., "Organic Thin-Film Transistors Based on Carbonyl-Functionalized Quaterthiophenes: High Mobility N-Channel Semi-conductors and Ambipolar Transport," J. Am. Chem. Soc. vol. 127, pp. 1348-1349, (2005).

Zotti et al., "Novel, Highly Conducting, and Soluble Polymers from Anodic Coupling of Alkyl-Substituted Cyclopentadithiophene Monomers," Macromolecules, 27:1938-1942, 1994.

Zotti et al., "Polymers, Dimers and Radical Cations from Electrochemical Oxidation of Interring-Bridged Thiophene and Thiophene-Phenylene Tetramers," Macromolecular Chemistry and Physics, 202:3049-3056, 2001.

Bjorstrom et al., "Multilayer formation in spin-coated thin films of low-bandgap polyfluorene: PCBM blends," J. Phys. Condens. Matter 17:L529-L534, 2005.

Drechsel et al., "High efficiency organic solar cells based on single or multiple PIN structures," Thin Solid Films, vol. 451-452:515-517, 2004.

Greenham et al., "Charge separation and transport in conjugated polymer/cadmium selenide nanocrystal composites studied by photoluminescence quenching and photoconductivity," Synthetic Metals, 84:545-546, 1997.

* cited by examiner

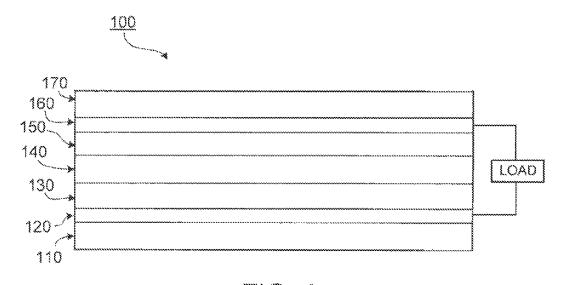


FIG. 1

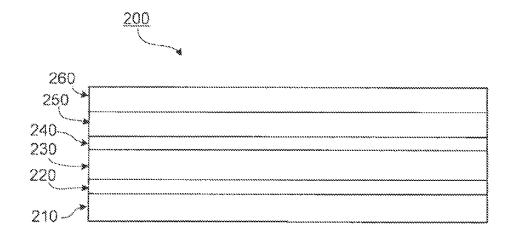


FIG. 2

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PHOTOVOLTAIC CELL WITH THIAZOLE-CONTAINING POLYMER

This application is a divisional and claims benefit of U.S. application Ser. No. 13/191,264, filed Jul. 26, 2011, which is a divisional and claims benefit of U.S. application Ser. No. 11/851,591, filed Sep. 7, 2007, now U.S. Pat. No. 8,008,424, which in turn claims benefit of U.S. Provisional Patent Application Ser. No. 60/850,845, filed Oct. 11, 2006. The entire contents of each of the applications are hereby incorporated by reference.

STATEMENT REGARDING FEDERALLY SPONSORED RESEARCH

This invention was supported by contract number W911QY-04-C-0070 awarded by the Defense Advanced Research Projects Agency. The U.S. Government has certain rights in the invention.

TECHNICAL FIELD

This invention relates to photovoltaic cells with thiazole-containing polymers, as well as related components; systems, 25 and methods.

BACKGROUND

Photovoltaic cells are commonly used to transfer energy in 30 the form of light into energy in the form of electricity. A typical photovoltaic cell includes a photoactive material disposed between two electrodes. Generally, light passes through one or both of the electrodes to interact with the photoactive material. As a result, the ability of one or both of 35 the electrodes to transmit light (e.g., light at one or more wavelengths absorbed by a photoactive material) can limit the overall efficiency of a photovoltaic cell. In many photovoltaic cells, a film of semiconductive material (e.g., indium tin oxide) is used to form the electrode(s) through which light 40 passes because, although the semiconductive material can have a lower electrical conductivity than electrically conductive materials, the semiconductive materials.

SUMMARY

This invention relates to photovoltaic cells with thiazole-containing polymers (e.g., polymers containing a bithiazole, cyclopentadithiazole, or thiazolothiazole moiety), as well as 50 related components, systems, and methods.

An aspect of the invention relates to new combinations of monomers for preparing polymers, which have properties suitable for use as charge carriers in the active layer of a photovoltaic cell.

In one aspect, this invention features a polymer including a first comonomer repeat unit comprising a cyclopentadithiazole moiety; and a second comonomer repeat unit different from the first comonomer repeat unit.

In another aspect, this invention features a polymer including a first comonomer repeat unit comprising a thiazolothiazole moiety; and a second comonomer repeat unit different from the first comonomer repeat unit. The second comonomer repeat unit is not a phenyl moiety or a fluorene moiety.

In another aspect, this invention features a polymer including a first comonomer repeat unit comprising a thiazole moiety; and a second comonomer repeat unit different from the

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first comonomer repeat unit. The second comonomer repeat unit is not a thiophene moiety or a fluorene moiety.

In still another aspect, this invention features an article that includes a first electrode, a second electrode, and a photoactive material disposed between the first and second electrodes. The photoactive material includes a polymer described above. The article is configured as a photovoltaic cell

Embodiments can include one or more of the following features.

In some embodiments, the first comonomer repeat unit includes a cyclopentadithiazole moiety of formula (1):

$$(1)$$

$$N$$

$$R_1$$

$$R_2$$

in which each of R_1 and R_2 , independently, is $H, C_1\text{-}C_{20}$ alkyl, $C_1\text{-}C_{20}$ alkoxy, $C_3\text{-}C_{20}$ cycloalkyl, $C_1\text{-}C_{20}$ heterocycloalkyl, aryl, heteroaryl, halo, CN, OR, C(O)R, C(O)OR, or SO_2R ; R being $H,\ C_1\text{-}C_{20}$ alkyl, $C_1\text{-}C_{20}$ alkoxy, aryl, heteroaryl, $C_3\text{-}C_{20}$ cycloalkyl, or $C_1\text{-}C_{20}$ heterocycloalkyl. For example, each of R_1 and R_2 , independently, can be $C_1\text{-}C_{20}$ alkoxy or $C_1\text{-}C_{20}$ alkyl optionally substituted with $C_1\text{-}C_{20}$ alkoxy or halo

In some embodiments, the first comonomer repeat unit comprises a thiazole moiety of formula (23):

in which R_5 is H, C_1 - C_{20} alkyl, C_1 - C_{20} alkoxy, C_3 - C_{20} cycloalkyl, C_1 - C_{20} heterocycloalkyl, aryl, heteroaryl, halo, CN, OR, C(O)R, C(O)OR, or SO_2R ; R being H, C_1 - C_{20} alkyl, C_1 - C_{20} alkoxy, aryl, heteroaryl, C_3 - C_{20} cycloalkyl, or C_1 - C_{20} heterocycloalkyl. For example, R_5 can be hexyl.

In some embodiments, the first comonomer repeat unit comprises a thiazolothiazole moiety of formula (25):

In some embodiments, the second comonomer repeat unit includes a silacyclopentadithiophene moiety, a benzothiadiazole moiety, a thiadiazoloquinoxaline moiety, a cyclopentadithiophene moiety, a cyclopentadithiophene oxide moiety, a benzothiazole moiety, a benzothiazole moiety, a thienothiophene oxide moiety, a thienothiophene moiety, a thienothiophene oxide moiety, a dithienothiophene moiety, a dithienothiophene oxide moiety, a fluorene moiety, a fluorene moiety, a fluorene moiety, a silole moiety, a thiazole moiety, a selenophene moiety, a silole moiety, a thiazole moiety, a selenophene moiety, a silole moiety, a thiazole moiety, a selenophene moiety, a silole moiety, a thiazole moiety.

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ety, a cyclopentadithiazole moiety, a naphthothiadiazole moiety, a thienopyrazine moiety, a thiophene moiety, an oxazole moiety, an imidazole moiety, a pyrimidine moiety, a benzoxazole moiety, or a benzimidazole moiety.

In some embodiments, the second comonomer repeat unit includes a cyclopentadithiazole moiety of formula (1), a benzothiadiazole moiety of formula (2), a thiadiazoloquinoxaline moiety of formula (3), a cyclopentadithiophene dioxide moiety of formula (4), a cyclopentadithiophene monoxide $_{10}$ moiety of formula (5), a benzoisothiazole moiety of formula (6), a benzothiazole moiety of formula (7), a thiophene dioxide moiety of formula (8), a cyclopentadithiophene dioxide moiety of formula (9), a cyclopentadithiophene tetraoxide moiety of formula (10), a thienothiophene moiety of formula 15 (11), a thienothiophene tetraoxide moiety of formula (12), a dithienothiophene moiety of formula dithienothiophene dioxide moiety of formula (14), a dithienothiophene tetraoxide moiety of formula (15), a tetrahydroisoindole moiety of formula (16), a thienothiophene 20 dioxide moiety of formula (17), a dithienothiophene dioxide moiety of formula (18), a fluorene moiety of formula (19), a silole moiety of formula (20), a cyclopentadithiophene moiety of formula (21), a fluorenone moiety of formula (22), a thiazole moiety of formula (23), a selenophene moiety of formula (24), a thiazolothiazole moiety of formula (25), a naphthothiadiazole moiety of formula (26), a thienopyrazine moiety of formula (27), a silacyclopentadithiophene moiety of formula (28), a thiophene moiety of formula (29), an 30 oxazole moiety of formula (30), an imidazole moiety of formula (31), a pyrimidine moiety of formula (32), a benzoxazole moiety of formula (33), or a benzimidazole moiety of formula (34):

$$\begin{array}{c} (1) \\ N \\ N \\ R_1 \\ R_2 \end{array}$$

$$R_5$$
 R_6 (2) 45

$$R_{5}$$
 R_{6} R_{6}

$$R_5$$
 R_7 R_7 R_6 R_6

$$R_7$$
 R_7
 R_8
 R_6
 R_6

$$R_{5}$$
 R_{6} R_{6} R_{6}

$$(9)$$

$$R_{5}$$

$$R_{6}$$

$$(10)$$

$$R_{5}$$

$$R_{6}$$

$$R_5$$
 (11)

$$R_5$$
 R_6 R_6

$$R_5$$
 R_6 R_6

$$R_5$$
 R_6
 R_6
 R_6
 R_6
 R_6
 R_6

$$\begin{array}{c}
R_5 \\
S \\
S \\
S
\end{array}$$

$$\begin{array}{c}
R_6 \\
S \\
S
\end{array}$$

$$\begin{array}{c}
S \\
S \\
S
\end{array}$$

$$\begin{array}{c}
S \\
S \\
S
\end{array}$$

$$\begin{array}{c}
S \\
S \\
S
\end{array}$$

$$R_5$$
 R_6 (19)

$$R_7$$
 R_8 (20) 60 R_5 R_6

$$R_5$$
 R_6 (24)

$$R_5$$
 R_6 (27)

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & & \\ R_7 & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

$$\begin{array}{c}
(29) \\
R_5 \\
R_6
\end{array}$$

(32)

(33)

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$$N \longrightarrow R_5$$
 $N \longrightarrow R_6$

$$R_5$$
 R_6 or

in which each of X and Y, independently, is CH2, O, or S; each of R_5 and R_6 , independently, is H, $C_1\text{-}C_{20}$ alkyl, $C_1\text{-}C_{20}$ alkoxy, C_3 - C_{20} cycloalkyl, C_1 - C_{20} heterocycloalkyl, aryl, heteroaryl, halo, CN, OR, C(O)R, C(O)OR, or SO₂R, in which R is H, C₁-C₂₀ alkyl, C₁-C₂₀ alkoxy, aryl, heteroaryl, 55 C_3 - C_{20} cycloalkyl, or C_1 - C_{20} heterocycloalkyl; and each of $\rm R_7$ and $\rm R_8,$ independently, is H, $\rm C_1\text{-}C_{20}$ alkyl, $\rm C_1\text{-}C_{20}$ alkoxy, aryl, heteroaryl, C₃-C₂₀ cycloalkyl, or C₃-C₂₀ heterocy-

In some embodiments, the polymer further includes a third 60 comonomer repeat unit different from the first and second comonomer repeat units. The third comonomer repeat unit can include a thiophene moiety (e.g., a unsubstituted or substituted thiophene moiety).

In some embodiments, the polymer can be either an electron donor material or an electron acceptor material.

in which n can be an integer greater than 1.

In some embodiments, the photovoltaic cell can be a tandem photovoltaic cell.

In some embodiments, the photoactive material can include an electron acceptor material. In some embodiments, the electron acceptor material can be a fullerene (e.g., C61-20 phenyl-butyric acid methyl ester, PCBM).

In some embodiments, the polymer and the electron acceptor material each can have a LUMO energy level. The LUMO energy level of the polymer can be at least about 0.2 eV (e.g., at least about 0.3 eV) less negative than the LUMO energy 25 level of the electron acceptor material.

In some embodiments, the device can be an organic semiconductive device. In certain embodiments, the device can be a member selected from the group consisting of field effect transistors, photodetectors, photovoltaic detectors, imaging 30 devices, light emitting diodes, lasing devices, conversion layers, amplifiers and emitters, storage elements, and electrochromic devices.

Embodiments can provide one or more of the following advantages.

In some embodiments, using a polymer containing a thiazole moiety can be advantageous because the thiazole moiety can contribute to a shift in the maximum absorption wavelength toward the red or near IR region of the electromagnetic spectrum. When such a polymer is incorporated into a pho-40 tovoltaic cell, the current and efficiency of the cell can increase.

In some embodiments, substituted fullerenes or polymers containing substituted monomer repeat units (e.g., substituted with long-chain alkoxy groups such as oligomeric eth-45 ylene oxides or fluorinated alkoxy groups) can have improved solubility in organic solvents and can form a photoactive layer with improved morphology.

In some embodiments, a polymer containing a thiazole moiety can absorb light at a relatively long wavelength and have improved solubility in organic solvents. In some embodiments, a polymer containing a thiazole moiety can be used to prepare an electron donor material with improved semiconductive properties.

In some embodiments, a photovoltaic cell containing a polymer described above can have a band gap that is relatively ideal for its intended purposes.

In some embodiments, a photovoltaic cell having high cell voltage can be created, whereby the HOMO level of the polymer is at least about 0.2 electron volts more negative relative to the LUMO or conduction band of an electron acceptor material.

In some embodiments, a photovoltaic cell containing a polymer described above can have relatively fast and efficient transfer of an electron to an electron acceptor material, whereby the LUMO of the donor is at least about 0.2 electron volt (e.g., at least about 0.3 electron volt) less negative than the conduction band of the electron acceptor material.

In some embodiments, a photovoltaic cell containing a polymer described above can have relatively fast charge separation, whereby the charge mobility of the positive charge, or hole, is relatively high and falls within the range of 10^{-4} to 10^{-1} cm²/Vs.

In some embodiments, the polymer is soluble in an organic solvent and/or film forming.

In some embodiments, the polymer is optically non-scattering.

In some embodiments, the polymer can be used in organic $\ ^{10}$ field effect transistors and OLEDs.

Other features and advantages of the invention will be apparent from the description, drawings, and claims.

DESCRIPTION OF DRAWINGS

FIG. 1 is a cross-sectional view of an embodiment of a photovoltaic cell.

FIG. 2 is a schematic of a system containing one electrode between two photoactive layers.

Like reference symbols in the various drawings indicate like elements.

DETAILED DESCRIPTION

FIG. 1 shows a cross-sectional view of a photovoltaic cell 100 that includes a substrate 110, a cathode 120, a hole carrier layer 130, an active layer 140 (containing an electron acceptor material and an electron donor material), a hole blocking layer 150, an anode 160, and a substrate 170.

In general, during use, light impinges on the surface of substrate 110, and passes through substrate 110, cathode 120, and hole carrier layer 130. The light then interacts with active layer 140, causing electrons to be transferred from the electron donor material (e.g., a polymer described above) to the 35 electron acceptor material (e.g., PCBM). The electron acceptor material then transmits the electrons through hole blocking layer 150 to anode 160, and the electron donor material transfers holes through hole carrier layer 130 to cathode 120. Anode 160 and cathode 120 are in electrical connection via an 40 external load so that electrons pass from anode 160, through the load, and to cathode 120.

Electron acceptor materials of active layer **140** can include fullerenes. In some embodiments, active layer **140** can include one or more unsubstituted fullerenes and/or one or 45 more substituted fullerenes. Examples of unsubstituted fullerenes include C_{60} , C_{70} , C_{76} , C_{78} , C_{82} , C_{84} , and C_{92} . Examples of substituted fullerenes include PCBM or fullerenes substituted with C_1 - C_{20} alkoxy optionally further substituted with C_1 - C_{20} alkoxy or halo (e.g., $(OCH_2CH_2)_2$ 50 OCH $_3$ or $OCH_2CF_2OCF_2CF_2OCF_3$). Without wishing to be bound by theory, it is believed that fullerenes substituted with long-chain alkoxy groups (e.g., oligomeric ethylene oxides) or fluorinated alkoxy groups have improved solubility in organic solvents and can form a photoactive layer with 55 improved morphology.

In some embodiments, the electron acceptor materials can include polymers (e.g., homopolymers or copolymers). A polymers mentioned herein include at least two identical or different monomer repeat units (e.g., at least 5 monomer 60 repeat units, at least 10 monomer repeat units, at least 50 monomer repeat units, at least 100 monomer repeat units, or at least 500 monomer repeat units). A copolymer mentioned herein refers to a polymer that includes at least two co-monomers of differing structures. In some embodiments, the polymers used as an electron acceptor material can include one or more monomer repeat units listed in Tables 1 and 2 below.

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Specifically, Table 1 lists examples of electron donating monomer repeat units that can serve as a conjugative link. Table 2 lists examples of electron withdrawing monomer repeat units. Note that depending on the substituents, monomer repeat units listed in Table 1 can be electron withdrawing and monomer repeat units listed in Table 2 can also be electron donating. Preferably, the polymers used as an electron acceptor material include a high molar percentage (e.g., at least about 50%, at least about 60%, at least about 70%, at least about 80%, at least about 90%) of an electron withdrawing monomer repeat unit.

Electron donor materials of active layer 140 can include polymers (e.g., homopolymers or copolymers). In some 15 embodiments, the polymers used as an electron donor material can include one or more monomer repeat units listed in Tables 1 and 2. Preferably, the polymers used as an electron donor material include a high molar percentage (e.g., at least about 50%, at least about 60%, at least about 70%, at least about 80%, at least about 90%) of an electron donating monomer repeat unit. In some embodiments, the polymers include a monomer repeat unit containing C1-C20 alkoxy on a ring, which is optionally further substituted with C_1 - C_{20} alkoxy or halo (e.g., (OCH₂CH₂)₂OCH₃ or OCH₂CF₂OCF₂CF₂OCF₃). Without wishing to be bound by theory, it is believed that polymers containing monomer repeat units substituted with long-chain alkoxy groups (e.g., oligomeric ethylene oxides) or fluorinated alkoxy groups have improved solubility in organic solvents and can form an photoactive layer with improved morphology.

TABLE 1

$$R_7$$
 R_5
 R_6
 R_8
 R_8

TABLE 1-continued

TABLE 2-continued

Referring to formulas listed in Tables 1 and 2 above, each of X and Y, independently, can be CH $_2$, O, or S; each of R $_1$, R $_2$, R $_5$, and R $_6$, independently, can be H, C $_1$ -C $_{20}$ alkyl (e.g., branched alkyl or perflorinated alkyl), C $_1$ -C $_{20}$ alkoxy, C $_3$ -C $_{20}$ cycloalkyl, C $_1$ -C $_{20}$ heterocycloalkyl, aryl (e.g., phenyl or substituted phenyl), heteroaryl, halo, CN, OR, C(O)R, C(O)OR, or SO $_2$ R; R being H, C $_1$ -C $_{20}$ alkyl, C $_1$ -C $_{20}$ alkoxy, aryl, heteroaryl, C $_3$ -C $_{20}$ cycloalkyl, or C $_1$ -C $_{20}$ heterocycloalkyl; and each of R $_7$ and R $_8$, independently, is H, C $_1$ -C $_{20}$ alkyl, C $_1$ -C $_{20}$ alkoxy, aryl, heterocycloalkyl.

An alkyl can be saturated or unsaturated and branch or straight chained. A C₁-C₂₀ alkyl contains 1 to 20 carbon atoms (e.g., one, two, three, four, five, six, seven, eight, nine, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, and 20 carbon atoms). Examples of alkyl moieties include -CH₃, -CH₂-, -CH₂=CH₂-, -CH₂-CH=CH₂, and branched 30 —C₃H₇. An alkoxy can be branch or straight chained and saturated or unsaturated. A C₁-C₂₀ alkoxy contains an oxygen radical and 1 to 20 carbon atoms (e.g., one, two, three, four, five, six, seven, eight, nine, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, and 20 carbon atoms). Examples of alkoxy moieties 35 include —OCH₃ and —OCH—CH—CH₃. A cycloalkyl can be either saturated or unsaturated. A C₃-C₂₀ cycloalkyl contains 3 to 20 carbon atoms (e.g., three, four, five, six, seven, eight, nine, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, and 20 carbon atoms). Examples of cycloalkyl moieities include cyclohexyl 40 and cyclohexen-3-yl. A heterocycloalkyl can also be either saturated or unsaturated. A C₃-C₂₀ heterocycloalkyl contains at least one ring heteroatom (e.g., O, N, and S) and 3 to 20 carbon atoms (e.g., three, four, five, six, seven, eight, nine, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, and 20 carbon atoms). 45 Examples of heterocycloalkyl moieties include 4-tetrahydropyranyl and 4-pyranyl. An aryl can contain one or more aromatic rings. Examples of aryl moieties include phenyl, phenylene, naphthyl, naphthylene, pyrenyl, anthryl, and phenanthryl. A heteroaryl can contain one or more aromatic 50 rings, at least one of which contains at least one ring heteroatom (e.g., O, N, and S). Examples of heteroaryl moieties include furyl, furylene, fluorenyl, pyrrolyl, thienyl, oxazolyl, imidazolyl, thiazolyl, pyridyl, pyrimidinyl, quinazolinyl, quinolyl, isoquinolyl, and indolyl.

Alkyl, alkoxy, cycloalkyl, heterocycloalkyl, aryl, and heteroaryl mentioned herein include both substituted and unsubstituted moieties, unless specified otherwise. Examples of substituents on cycloalkyl, heterocycloalkyl, aryl, and heteroaryl include C₁-C₂₀ alkyl, C₃-C₂₀ cycloalkyl, C₁-C₂₀
alkoxy, aryl, aryloxy, heteroaryl, heteroaryloxy, amino, C₁-C₁₀ alkylamino, C₁-C₂₀ dialkylamino, arylamino, diarylamino, hydroxyl, halogen, thio, C₁-C₁₀ alkylthio, arylthio, C₁-C₁₀ alkylsulfonyl, arylsulfonyl, cyano, nitro, acyl, acyloxy, carboxyl, and carboxylic ester. Examples of substituents on alkyl include all of the above-recited substituents except C₁-C₂₀ alkyl. Cycloalkyl, heterocycloalkyl, aryl, and heteroaryl also include fused groups.

X R_7 N Y R_5 R_6

$$\begin{pmatrix} R_5 \\ \\ \\ \\ \\ \\ \end{pmatrix}$$
 $\begin{pmatrix} R_6 \\ \\ \\ \\ \\ \end{pmatrix}$ $\begin{pmatrix} R_6 \\ \\ \\ \\ \\ \end{pmatrix}$

$$R_{5}$$

$$\underbrace{\left\langle \left\langle \right\rangle \right\rangle }_{N}\underbrace{\left\langle \left\langle \right\rangle \right\rangle }_{S}$$

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The monomers for preparing the polymers mentioned herein may contain a non-aromatic double bond and one or more asymmetric centers. Thus, they can occur as racemates and racemic mixtures, single enantiomers, individual diastereomers, diastereomeric mixtures, and cis- or trans-isomeric forms. All such isomeric forms are contemplated.

The copolymers described above can be prepared by methods known in the art. For example, a copolymer can be prepared by a cross-coupling reaction between one or more $_{10}$ comonomers containing two alkylstannyl groups and one or more comonomers containing two halo groups in the presence of a transition metal catalyst. As another example, a copolymer can be prepared by a cross-coupling reaction between one or more comonomers containing two borate 15 groups and one or more comonomers containing two halo groups in the presence of a transition metal catalyst. The comonomers can be prepared by the methods described herein or by the methods know in the art, such as those described in U.S. patent application Ser. No. 11/486,536, 20 Coppo et al., Macromolecules 2003, 36, 2705-2711 and Kurt et al., J. Heterocycl. Chem. 1970, 6, 629, the contents of which are hereby incorporated by reference.

The monomers for preparing the copolymers described above are either commercially available or can be prepared by methods known in the art. For example, a cyclopentadithiazole monomer can be prepared by the method illustrated in Scheme I below. The cyclopentadithiazole monomer thus obtained can then react with another monomer (such as those shown in Scheme II below) to form a copolymer described above.

$$\begin{array}{c} \underline{Scheme\ II} \\ \\ \underline{N} \\ \\ \underline{N}$$

Table 3 below lists an exemplary polymer (i.e., polymer 1) described in the Summary section above. This polymer can have unique properties, which make it particularly suitable as a charge carrier in the active layer of a photovoltaic cell. For example, an organic photovoltaic cell include this polymer in the active layer can have an efficiency more than 3%. Polymer 1 can be obtained by the methods described in Example 2 below.

Generally, one co-monomer in the polymers described in the Summary section above contains a thiazole moiety (e.g., a bithiazole, thiazolothiazole, or cyclopentadithiazole moiety). An advantage of a co-polymer containing a thiazole moiety is that its absorption wavelength can shift toward the red and near IR portion (e.g., 650-800 nm) of the electromagnetic spectrum, which is not accessible by most other polymers. When such a co-polymer is incorporated into a photovoltaic cell, it enables the cell to absorb the light in this region 10 of the spectrum, thereby increasing the current and efficiency of the cell.

The polymers described above can be useful in solar power technology because the band gap is close to ideal for a photovoltaic cell (e.g., a polymer-fullerene cell). The HOMO level of the polymers can be positioned correctly relative to the LUMO of an electron acceptor (e.g., PCBM) in a photovoltaic cell (e.g., a polymer-fullerene cell), allowing for high cell voltage. The LUMO of the polymers can be positioned $\,^{20}$ correctly relative to the conduction band of the electron acceptor in a photovoltaic cell, thereby creating efficient transfer of an electron to the electron acceptor. For example, using a polymer having a band gap of about 1.4-1.6 eV can 25 significantly enhance cell voltage. Cell performance, specifically efficiency, can benefit from both an increase in photocurrent and an increase in cell voltage, and can approach and even exceed 15% efficiency. The positive charge mobility of 30 the polymers can be relatively high and approximately in the range of 10^{-4} to 10^{-1} cm²/Vs. In general, the relatively high positive charge mobility allows for relatively fast charge solvent and/or film forming. Further, the polymers can be optically non-scattering.

Components in photovoltaic cell other than the electron acceptor materials and the electron donor materials are known in the art, such as those described in U.S. patent application Ser. No. 10/723,554, the contents of which are incorporated herein by references.

In some embodiments, the polymer described above can be used as an electron donor material or an electro acceptor 45 material in a system in which two photovoltaic cells share a common electrode. Such a system is also known as tandem photovoltaic cell. Examples of tandem photovoltaic cells are discussed in U.S. patent application Ser. No. 10/558,878, filed Nov. 29, 2005, the contents of which are hereby incorporated by reference.

As an example, FIG. 2 is a schematic of a tandem photovoltaic cell 200 having a substrate 210, three electrodes 220, **240**, and **260**, and two photoactive layers **230** and **250**. Elec- 55 trode 240 is shared between photoactive layers 230 and 250, and is electrically connected with electrodes 220 and 260. In general, electrodes 220, 240, and 260 can be formed of an electrically conductive material, such as those described in U.S. patent application Ser. No. 10/723,554. In some embodiments, one or more (i.e., one, two, or three) electrodes 220, 240, and 260 is a mesh electrode. In some embodiments, one or more electrodes 220, 240, and 260 is formed of a semiconductive material. Examples of semiconductive materials 65 include titanium oxides, indium tin oxides, fluorinated tin oxides, tin oxides, and zinc oxides. In certain embodiments,

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one or more (i.e., one, two, or three) electrodes 220, 240, and 260 are formed of titanium dioxide. Titanium dioxide used to prepare an electrode can be in any suitable forms. For example, titanium dioxide can be in the form of interconnected nanoparticles. Examples of interconnected titanium dioxide nanoparticles are described, for example, in U.S. Pat. No. 7,022,910, the contents of which are incorporated herein by reference. In some embodiments, at least one (e.g., one, two, or three) of electrodes 220, 240, and 260 is a transparent electrode. As referred to herein, a transparent electrode is formed of a material which, at the thickness used in a photovoltaic cell, transmits at least about 60% (e.g., at least about 70%, at least about 75%, at least about 80%, at least about 85%, at least about 90%, at least about 95%) of incident light at a wavelength or a range of wavelengths used during operation of the photovoltaic cell. In certain embodiments, both electrodes 220 and 260 are transparent electrodes.

Each of photoactive layers 230 and 250 can contain at least one semiconductive material. In some embodiments, the semiconductive material in photoactive layer 230 has the same band gap as the semiconductive material in photoactive layer 250. In certain embodiments, the semiconductive material in photoactive layer 230 has a band gap different from that of the semiconductive material in photoactive layer 250. Without wishing to be bound by theory, it is believed that incident light not absorbed by one photoactive layer can be absorbed by the other photoactive layer, thereby maximizing the absorption of the incident light.

In some embodiments, at least one of photoactive layers separation. The polymers can also be soluble in an organic 35 230 and 250 can contain an electron acceptor material (e.g., PCBM or a polymer described above) and an electron donor material (e.g., a polymer described above). In general, suitable electron acceptor materials and electron donor materials 40 can be those described above. In certain embodiments, each of photoactive layers 230 and 250 contains an electron acceptor material and an electron donor material.

> Substrate 210 can be formed of one or more suitable polymers, such as those described in U.S. patent application Ser. No. 10/723,554. In some embodiments, an additional substrate (not shown in FIG. 2) can be disposed on electrode 260.

> Photovoltaic cell 200 can further contain a hole carrier layer (not shown in FIG. 2) and a hole blocking layer (not shown in FIG. 2), such as those described in U.S. patent application Ser. No. 10/723,554.

> While photovoltaic cells have been described above, in some embodiments, the polymers described herein can be used in other devices and systems. For example, the polymers can be used in suitable organic semiconductive devices, such as field effect transistors, photodetectors (e.g., IR detectors), photovoltaic detectors, imaging devices (e.g., RGB imaging devices for cameras or medical imaging systems), light emitting diodes (LEDs) (e.g., organic LEDs or IR or near IR LEDs), lasing devices, conversion layers (e.g., layers that convert visible emission into IR emission), amplifiers and emitters for telecommunication (e.g., dopants for fibers), storage elements (e.g., holographic storage elements), and electrochromic devices (e.g., electrochromic displays).

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The following examples are illustrative and not intended to be limiting.

EXAMPLE 1

Preparation of 2,6-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-4H-4,4-bis(2'-ethylhexyl)cyclopenta[2,1-b:3,4-b']thiophene

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argon three times. To this flask was then added 20 mL of dry, distilled THF. The resulting solution was cooled to -78° C. and 4.35 mL (10.88 mmol, 4 equv.) of 2.5M BuLi was added dropwise. The reaction was stirred for 1 hout at -78° C. and then warmed to room temperature and stirred for an additional 3 hours. The solution was cooled again to -78° C. and 2.77 mL (13.6 mmol, 5 equiv.) of 2-isopropoxy-4.4.5.5-tetramethyl-1,3,2-dioxaborolane was added in one portion via syringe. The reaction was stirred at -78° C. for 1 hour and then allowed to warm to room temperature overnight. The solution was poured into water and extracted with 4×150 mL of methyl tert-butyl ether. The organic layers were combined and washed with 2×150 mL of brine, dried with anhydrous MgSO₄, and filtered. The solvent was removed under vacuum to yield and orange oil, which was purified by column chromatography (5% EtOAc in hexanes) to yield a colorless, viscous oil, 1.34 g (75% yield).

EXAMPLE 2

Preparation of Polymer 1

100 mL oven dried Schlenk flask was charged with 1.097 g (2.72 mmol) of 4H-4,4-bis(2'-ethylhexyl)cyclopenta[2,1-b:3, 4-b']dithiophene. The flask was evacuated and purged with

A 100 mL Schlenk flask was charged with 0.1515 g (0.231 mmol) of 2,6-bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2yl)-4H-4,4-bis(2'-ethylhexyl)cyclopenta[2,1-b:3,4-b'] 55 thiophene, 0.152 g (0.231 mmol) of 5,5'-bis(5-bromo-2-thienyl)-4,4'-dihexyl-2,2'-bithiazole, 2.1 mg Pd₂ dba₃ (0.00231 mmol), 4.2 mg PPh₃ (0.0162 mmol), and 35 mg (0.0855 mmol) of Aliquat 336. The flask, which was fitted with a condenser, was then evacuated and refilled with argon three times. The reagents were dissolved in a mixture of 20 mL of THF and 15 mL of toluene. 2 mL of a 2M Na₂CO₃ aquesous solution was added to the above solution while stirring. The reaction was heated at 90° C. for 3 days. A 1 mL THF solution of 14 mg (0.1155 mmol) of phenylboronic acid and 1.6 mg (0.00231 mmol) of PdCl₂(PPh₃)₂ was added. Heating was continued for an additional 24 hours. After the reaction was then cooled to 80° C., 10 mL of a 7.5% sodium diethyldithio-

carbamate solution in water was added. The mixture was heated at 80° C. with stirring for 18 hours. After the reaction was cooled, the organic layer was separated and washed with warm water (3×100 mL). The toluene solution was concentrated and then poured into 750 mL of stirring MeOH. After 5 the solution was filtered, the dark precipitate was collected and washed with MeOH. The precipitate was then transferred to a Soxhlet thimble and washed with acetone overnight. The product thus obtained was dried under vacuum to give 0.176 g of brown solid (0.195 mmol, 84%). ¹H NMR (200 MHz, 100 CDCl₃): 87.2-7.1 (br, 6H), 3.0 (m, 4H), 1.86 (m, 8H), 1.6 (br, 16H), 1.20-0.65 (br, 32H).

Other embodiments are in the claims.

What is claimed is:

- 1. An article, comprising:
- a first electrode;
- a second electrode; and
- a photoactive material disposed between the first and second electrodes, the photoactive material comprising a polymer including a first comonomer repeat unit and a 20 second comonomer repeat unit different from the first comonomer repeat unit;
- wherein the first comonomer repeat unit comprises a thiazole moiety; the second comonomer repeat unit is not a thiophene moiety or a fluorene moiety; and the article is 25 configured as a photovoltaic cell.
- 2. The article of claim 1, wherein the first comonomer repeat unit comprises a thiazole moiety of formula (23):

wherein R_5 is H, C_1 - C_{20} alkyl, C_1 - C_{20} alkoxy, C_3 - C_{20} cycloalkyl, C_1 - C_{20} heterocycloalkyl, aryl, heteroaryl, halo, CN, OR, C(O)R, C(O)OR, or SO_2R ; R being H, 40 C_1 - C_{20} alkyl, C_1 - C_{20} alkoxy, aryl, heteroaryl, C_3 - C_{20} cycloalkyl, or C_1 - C_{20} heterocycloalkyl.

- 3. The article of claim 2, wherein R_5 is C_1 - C_{20} alkoxy or C_1 - C_{20} alkyl optionally substituted with C_1 - C_{20} alkoxy or halo.
 - **4**. The article of claim **3**, wherein R_5 is hexyl.
- 5. The article of claim 1, wherein the second comonomer repeat unit comprises a silacyclopentadithiophene moiety, a benzothiadiazole moiety, a thiadiazoloquinoxaline moiety, a cyclopentadithiophene moiety, a cyclopentadithiophene moiety, a benzothiazole moiety, a benzoisothiazole moiety, a benzothiazole moiety, a thiophene oxide moiety, a thienothiophene moiety, a dithienothiophene oxide moiety, a dithienothiophene moiety, a dithienothiophene oxide moiety, a tetrahydroisoindole moiety, a fluorenone moiety, a thiazole moiety, a selenophene smoiety, a silole moiety, a thiazole moiety, a cyclopentadithiazole moiety, a naphthothiadiazole moiety, a thienopyrazine moiety, an oxazole moiety, an imidazole moiety, a pyrimidine moiety, a benzoxazole moiety, or a benzimidazole moiety.
- 6. The article of claim 5, wherein the second comonomer repeat unit comprises a cyclopentadithiazole moiety of formula (1), a benzothiadiazole moiety of formula (2), a thiadiazoloquinoxaline moiety of formula (3), a cyclopentadithiophene dioxide moiety of formula (4), a 65 cyclopentadithiophene monoxide moiety of formula (5), a benzoisothiazole moiety of formula (6), a benzothiazole moi-

ety of formula (7), a thiophene dioxide moiety of formula (8), a cyclopentadithiophene dioxide moiety of formula (9), a cyclopentadithiophene tetraoxide moiety of formula (10), a thienothiophene moiety of formula (11), a thienothiophene tetraoxide moiety of formula (12), a dithienothiophene moiety of formula (13), a dithienothiophene dioxide moiety of formula (14), a dithienothiophene tetraoxide moiety of formula (15), a tetrahydroisoindole moiety of formula (16), a thienothiophene dioxide moiety of formula (17), a dithienothiophene dioxide moiety of formula (18), a silole moiety of formula (20), a cyclopentadithiophene moiety of formula (21), a fluorenone moiety of formula (22), a thiazole moiety of formula (23), a selenophene moiety of formula (24), a thiazolothiazole moiety of formula (25), a naphthothiadiazole moiety of formula (26), a thienopyrazine moiety of formula (27), a silacyclopentadithiophene moiety of formula (28), an oxazole moiety of formula (30), an imidazole moiety of formula (31), a pyrimidine moiety of formula (32), a benzoxazole moiety of formula (33), or a benzimidazole moiety of formula (34):

$$\begin{array}{c}
N \\
R_{\epsilon}
\end{array}$$
(3)

$$(4)$$

$$R_1 \quad R_2$$

$$(5)$$

$$R_{5}$$

$$R_{6}$$

$$R_5$$
 R_6 10 (7) R_7 R_7

$$R_{5}$$
 R_{6}
 R_{6}
 R_{6}
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 R_{6}
 R_{6}
 R_{6}
 R_{6}

$$R_5$$
 R_6 (8)

$$(9)$$
 30 R_5 R_6 R_6

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$$R_5$$
 R_6 (11) 45 R_6 R_6 R_6

$$R_5$$
 R_6
 R_6
 R_6
 R_6

$$X \xrightarrow{R_7} Y$$

$$X \xrightarrow{N} Y$$

$$R_5 \qquad R_6$$

$$R_6$$

$$\begin{array}{c}
R_5 \\
\hline
\\
S \\
\hline
\\
R_6
\end{array},$$
(17)

$$\begin{array}{c}
R_5 \\
S \\
S \\
S
\end{array}$$
(18)

$$\begin{array}{c}
R_7 \\
R_8 \\
R_5
\end{array}, \qquad (20)$$

$$R_7$$
 R_5 R_6 R_8 R_8 R_8

$$\stackrel{S}{\longrightarrow}$$
, (23)

$$R_5$$
 R_6
 R_5
 R_6
 R_7
 R_7

__

wherein

(26)

15

20

(28)

(30)

each of X and Y, independently, is CH₂, O, or S; each of R₁, R₂, R₅ and R₆, independently, is H, C₁-C₂₀ alkyl, C₁-C₂₀ alkoxy, C₃-C₂₀ cycloalkyl, C₁-C₂₀ heterocycloalkyl, aryl, heteroaryl, halo, CN, OR, C(O)R, C(O) OR, or SO₂R, in which R is H, C₁-C₂₀ alkyl, C₁-C₂₀ alkoxy, aryl, heteroaryl, C₃-C₂₀ cycloalkyl, or C₁-C₂₀ heterocycloalkyl; and

each of R_7 and R_8 , independently, is H, C_1 - C_{20} alkyl, C_1 - C_{20} alkoxy, aryl, heteroaryl, C_3 - C_{20} cycloalkyl, or C_3 - C_{20} heterocycloalkyl.

7. The article of claim 6, wherein the second comonomer repeat unit comprises the cyclopentadithiazole moiety of formula (1), the cyclopentadithiophene moiety of formula (21), or the silacyclopentadithiophene moiety of formula (28).

8. The article of claim **1**, further comprising a third comomoner repeat unit different from the first and second comonomer repeat units.

9. The article of claim 8, wherein the third commoner repeat unit comprises a thiophene moiety.

10. A polymer, comprising:

a first comonomer repeat unit comprising a thiazole moiety; and

a second comonomer repeat unit different from the first comonomer repeat unit, the second comonomer repeat unit not being a thiophene moiety or a fluorene moiety.

11. The polymer of claim 10, wherein the first comonomer repeat unit comprises a thiazole moiety of formula (23):

$$(31) 45 \qquad \qquad \begin{matrix} S \\ \\ \\ \\ \\ \\ \\ \\ \end{matrix} \end{matrix} , \qquad (23)$$

 $\begin{array}{c} \text{wherein } R_5 \text{ is } H, \ C_1\text{-}C_{20} \ \text{alkyl}, \ C_1\text{-}C_{20} \ \text{alkoxy}, \ C_3\text{-}C_{20} \\ \text{cycloalkyl}, \ C_1\text{-}C_{20} \ \text{heterocycloalkyl}, \ \text{aryl}, \ \text{heteroaryl}, \\ \text{halo, CN, OR, C(O)R, C(O)OR, or SO}_2R; \ R \ \text{being } H, \\ C_1\text{-}C_{20} \ \text{alkyl}, \ C_1\text{-}C_{20} \ \text{alkoxy, aryl}, \ \text{heteroaryl}, \ C_3\text{-}C_{20} \\ \text{cycloalkyl}, \ \text{or } C_1\text{-}C_{20} \ \text{heterocycloalkyl}. \end{array}$

12. The polymer of claim 11, wherein R_5 is C_1 - C_{20} alkoxy or C_1 - C_{20} alkyl optionally substituted with C_1 - C_{20} alkoxy or halo.

13. The polymer of claim 12, wherein R_5 is hexyl.

14. The polymer of claim 10, wherein the second comonomer repeat unit comprises a silacyclopentadithiophene moiety, a benzothiadiazole moiety, a thiadiazoloquinoxaline moiety, a cyclopentadithiophene moiety, a cyclopentadithiophene moiety, a benzothiazole moiety, a benzothiazole moiety, a thiophene oxide moiety, a thienothiophene moiety, a thienothiophene moiety, a dithienothiophene oxide moiety, a tetrahydroisoindole moiety, a fluorenone moiety, a thiazole moiety, a selenophene moiety, a silole moiety, a thiazolothia-

zole moiety, a cyclopentadithiazole moiety, a naphthothiadiazole moiety, a thienopyrazine moiety, an oxazole moiety, an imidazole moiety, a pyrimidine moiety, a benzoxazole moiety, or a benzimidazole moiety.

15. The polymer of claim 14, wherein the second comonomer repeat unit comprises a cyclopentadithiazole moiety of formula (1), a benzothiadiazole moiety of formula (2), a thiadiazoloquinoxaline moiety of formula (3), a cyclopentadithiophene dioxide moiety of formula (4), a cyclopenta- $_{10}$ dithiophene monoxide moiety of formula (5), a benzoisothiazole moiety of formula (6), a benzothiazole moiety of formula (7), a thiophene dioxide moiety of formula (8), a cyclopentadithiophene dioxide moiety of formula (9), a cyclopentadithiophene tetraoxide moiety of formula (10), a 15 thienothiophene moiety of formula (11), a thienothiophene tetraoxide moiety of formula (12), a dithienothiophene moiety of formula (13), a dithienothiophene dioxide moiety of formula (14), a dithienothiophene tetraoxide moiety of formula (15), a tetrahydroisoindole moiety of formula (16), a 20 thienothiophene dioxide moiety of formula (17), a dithienothiophene dioxide moiety of formula (18), a silole moiety of formula (20), a cyclopentadithiophene moiety of formula (21), a fluorenone moiety of formula (22), a thiazole moiety of formula (23), a selenophene moiety of formula (24), a thiazolothiazole moiety of formula (25), a naphthothiadiazole moiety of formula (26), a thienopyrazine moiety of formula (27), a silacyclopentadithiophene moiety of formula (28), an oxazole moiety of formula (30), an imidazole moiety of formula (31), a pyrimidine moiety of formula (32), a benzoxazole moiety of formula (33), or a benzimidazole moiety of formula (34):

$$R_5$$
 R_6 (2) 45

$$R_{5}$$
 R_{6}
 R_{6}

$$(4)$$

$$\downarrow S$$

$$R_5$$
 R_7 R_7 R_6 R_6 R_6

$$R_7$$
 R_7
 R_8
 R_6
 R_6

$$R_{5}$$
 R_{6}
 R_{6}
 R_{6}

$$(9)$$

$$R_{5}$$

$$R_{6}$$

$$(10)$$

$$R_{5}$$

$$R_{6}$$

$$R_{5}$$
 S
 R_{6}
 R_{6}
 R_{7}
 R_{8}

$$\begin{array}{c} R_5 \\ R_6 \end{array}$$

$$R_5$$
 R_6 R_6

$$\begin{array}{c} R_7 \\ X \\ N \\ Y \\ \end{array}$$

$$R_5$$
 R_6
 R_6
 R_6
 R_6
 R_6
 R_6

$$\begin{array}{c}
R_5 \\
\hline
\\
S
\end{array}$$

$$\begin{array}{c}
R_6 \\
\hline
\\
S
\end{array}$$

$$\begin{array}{c}
S
\end{array}$$

$$\begin{array}{c}
S
\end{array}$$

$$\begin{array}{c}
S
\end{array}$$

$$R_7$$
 R_8 (20)
$$R_5$$
 R_6

$$R_7$$
 R_8 R_8 R_8 R_8 R_8 R_8

$$\begin{array}{c}
\text{Se} \\
\text{Rs}
\end{array}$$

$$\begin{array}{c}
\text{Re} \\
\text{Re}
\end{array}$$

$$- \bigvee_{N}^{S} \bigvee_{S}^{N}, \qquad (25)$$

$$R_5$$
 R_6 (26)

$$R_5$$
 R_6 (27)

$$R_5$$
 (30)

$$\begin{array}{c}
H \\
N \\
N \\
R_5
\end{array},$$
(31)

(34)

-continued

$$R_5$$
 R_6
 R_6
 R_6
 R_6
 R_6
 R_6
 R_6
 R_6
 R_6

wherein

each of X and Y, independently, is CH₂, O, or S;

each of R₁, R₂, R₅ and R₆, independently, is H, C₁-C₂₀ alkyl, C₁-C₂₀ alkoxy, C₃-C₂₀ cycloalkyl, C₁-C₂₀ heterocycloalkyl, aryl, heteroaryl, halo, CN, OR, C(O)R, C(O) OR, or SO₂R, in which R is H, C₁-C₂₀ alkyl, C₁-C₂₀ alkoxy, aryl, heteroaryl, C₃-C₂₀ cycloalkyl, or C₁-C₂₀ heterocycloalkyl; and

each of R₇ and R₈, independently, is H, C₁-C₂₀ alkyl, C₁-C₂₀ alkoxy, aryl, heteroaryl, C₃-C₂₀ cycloalkyl, or C₃-C₂₀ heterocycloalkyl.

16. The polymer of claim 15, wherein the second comonomer repeat unit comprises the cyclopentadithiazole moiety of formula (1), the cyclopentadithiophene moiety of formula (21), or the silacyclopentadithiophene moiety of formula (28).

17. The polymer of claim 10, further comprising a third commoner repeat unit different from the first and second commoner repeat units.

18. The polymer of claim 17, wherein the third comomoner repeat unit comprises a thiophene moiety.

* * * *

UNITED STATES PATENT AND TRADEMARK OFFICE

CERTIFICATE OF CORRECTION

PATENT NO. : 9,123,895 B2 Page 1 of 1

APPLICATION NO. : 14/046567

DATED : September 1, 2015 INVENTOR(S) : Russell Gaudiana et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

In the specification:

Column 5, Line 5, Structure

Replace " With -- " With --
$$R_5$$
 R_6

In the claims:

Column 22, Line 50, Claim 6, Structure 4

Replace " With --
$$R_5$$
 R_6 R_6

Signed and Sealed this Fifth Day of July, 2016

Wichelle K. Lee

Michelle K. Lee

Director of the United States Patent and Trademark Office